

High-entropy rare-earth diborodcarbide: A novel class of high-entropy $(Y_{0.25}Yb_{0.25}Dy_{0.25}Er_{0.25})B_2C_2$ ceramics

Huidong Xu^{a,b}, Longfei Jiang^a, Ke Chen^{a,b}, Qing Huang^{a,b}, Xiaobing Zhou^{a,b,*}

^aEngineering Laboratory of Advanced Energy Materials, Ningbo Institute of Materials Technology and Engineering, Chinese Academy of Sciences, Ningbo 315201, China

^bUniversity of Chinese Academy of Sciences, Beijing 100049, China

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Abstract: A novel class of high-entropy rare-earth metal diborodcarbide $(Y_{0.25}Yb_{0.25}Dy_{0.25}Er_{0.25})B_2C_2$ (HE-REB₂C₂) ceramics was successfully fabricated using the *in-situ* reactive spark plasma sintering (SPS) technology for the first time. Single solid solution with a typical tetragonal structure was formed, having a homogeneous distribution of four rare-earth elements, such as Y, Yb, Dy, and Er. Coefficients of thermal expansion (CTEs) along the *a* and *c* directions (α_a and α_c) were determined to be 4.18 and 16.06 μK^{-1} , respectively. Thermal expansion anisotropy of the as-obtained HE-REB₂C₂ was attributed to anisotropy of the crystal structure of HE-REB₂C₂. The thermal conductivity (*k*) of HE-REB₂C₂ was $9.2 \pm 0.09 \text{ W} \cdot \text{m}^{-1} \cdot \text{K}^{-1}$, which was lower than that of YB₂C₂ ($19.2 \pm 0.07 \text{ W} \cdot \text{m}^{-1} \cdot \text{K}^{-1}$), DyB₂C₂ ($11.9 \pm 0.06 \text{ W} \cdot \text{m}^{-1} \cdot \text{K}^{-1}$), and ErB₂C₂ ($12.1 \pm 0.03 \text{ W} \cdot \text{m}^{-1} \cdot \text{K}^{-1}$), due to high-entropy effect and sluggish diffusion effect of high-entropy ceramics (HECs). Furthermore, Vickers hardness of HE-REB₂C₂ was slightly higher than that of REB₂C₂ owing to the solid solution hardening mechanism of HECs. Typical nano-laminated fracture morphologies, such as kink boundaries, delamination, and slipping were observed at the tip of Vickers indents, suggesting ductile behavior of HE-REB₂C₂. This newly investigated class of ductile HE-REB₂C₂ ceramics expanded the family of HECs to diboridcarbide compounds, which can lead to more research works on high-entropy rare-earth diboridcarbides in the near future.

Keywords: high-entropy rare-earth diboridcarbide; $(Y_{0.25}Yb_{0.25}Dy_{0.25}Er_{0.25})B_2C_2$ (REB₂C₂); high-entropy ceramics (HECs); thermal property; damage tolerance

1 Introduction

Ultra-high-temperature ceramics (UHTCs) have high melting points, outstanding oxidation resistance, along with excellent mechanical properties [1–4]. They have attracted increasing attention for aerospace applications,

such as leading edges and a nose cone of hypersonic vehicles, as well as surface coatings for SiC-based ceramic matrix composites [5–10]. Since the service temperatures (*T*) of UHTCs are usually above 1800 °C, most of UHTCs belong to highly covalent bonded ceramics, such as transition metal borides and carbides [11,12]. On the other hand, intrinsic brittleness and poor machinability of the transition metal borides and carbides obviously inhibited their applications as

* Corresponding author.

E-mail: zhouxb@nimte.ac.cn

ultra-high-temperature structure materials [13].

Rare-earth diborodiborides (REB_2C_2 , RE = Sc, Y, and lanthanide elements), which were first developed by Post *et al.* [14] in 1956, are a new group of ternary layered structure materials, similar to MAX phases (where M is an early transition metal, A is the IIIA, IVA, VA, or VIA group elements, and X is carbon (C) or nitrogen) [15]. YB_2C_2 is a typical compound of the REB_2C_2 family, which has been widely investigated [16–19]. The crystal structure of YB_2C_2 was reported as the tetragon with two types of a space group of $P4_2/mmc$ (No. 131) and $P4/mbm$ (No. 127) [20,21]. In YB_2C_2 , Y and B_2C_2 layers are alternatively stacking along the *c* direction. In the Y layers, Y–Y bonds are metallic bonds, while boron (B)–C bonds are covalent bonds in the B_2C_2 layers. The Y and B_2C_2 layers are mainly bonded by van der Waals force. Particularly, in the B_2C_2 layers, B and C atoms form four-member and eight-member B–C rings. The theoretical calculation results indicated that YB_2C_2 presents strong anisotropy in elastic properties owing to anisotropy of chemical bonds [19].

Furthermore, YB_2C_2 was reported as machinable layered ceramics with fracture toughness of $4.6 \pm 0.1 \text{ MPa} \cdot \text{m}^{1/2}$ and excellent damage tolerance. The residual flexural strength under Vickers contact load of 200 N was reported to be 80% [22]. Very recently, Li *et al.* [17] demonstrated that YB_2C_2 is stable at such a high temperature as 2500 °C for 20 min, which suggested that the melting point of YB_2C_2 is in the temperature range of 2500–2550 °C. Therefore, YB_2C_2 has been considered as a promising candidate for ultra-high-temperature applications.

On the other hand, high-entropy ceramics (HECs) are solid solution of inorganic compounds with one or more Wyckoff sites shared by equal or near equal atomic ratios of multi-principal elements [23]. HECs have drawn considerable attention due to their four core effects similar to high-entropy alloys (HEAs), i.e., (i) high-entropy effect: Configuration entropies in HEAs are much higher than those of conventional alloys, reducing free energy of a solid solution phase and promoting the formation of the solid solution [24]. (ii) Severe lattice distortion: The difference in the atomic size causes a lattice distortion, resulting in solid solution strengthening and hardening [25]. (iii) Sluggish diffusion effect inhibits the grain growth of a second phase [26]. (iv) Cocktail effect results in unexpected properties due to the solid solution formed by several

elements [27]. Compared to HEAs, HECs have larger space for improvement of the properties expanding through band structure and phonon engineering [23]. In contrast to conventional ceramics, HECs have been reported to possess excellent mechanical properties, such as superior-hardness, lower thermal conductivity (*k*), good thermodynamic stability, and improved corrosion and oxidation resistance [28–31]. Therefore, the high-entropy concept is a novel strategy for UHTCs due to low thermal conductivity and excellent thermal stability of such materials [28,32]. On the other hand, the hardness of the HECs is usually higher than their counterparts for traditional HE borides and carbides, resulting in solid solution strengthening and hardening mechanism. Therefore, the development of a ductile and damage tolerant HE ceramic for ultra-high-temperature applications is an interesting research topic.

Recently, the expansion of the HEC family has been relatively fast. A large number of papers on high-entropy carbides, borides, oxides, and silicides have been reported [33–36]. However, to the best of the authors' knowledge, there has been no report on the development of high-entropy rare-earth diboridecarbides.

Therefore, the aim of this work was to expand the family of HECs. A new class of ductile nature and damage tolerance high-entropy rare-earth diborodiboride with the composition of $(\text{Y}_{0.25}\text{Yb}_{0.25}\text{Dy}_{0.25}\text{Er}_{0.25})\text{B}_2\text{C}_2$ (REB_2C_2) was successfully synthesized using *in-situ* reactive spark plasma sintering (SPS) for the first time. Microstructures, coefficients of thermal expansion (CTEs) in the *a* and *c* directions (α_a and α_c), thermal conductivity, and Vickers hardness of the as-obtained high-entropy $(\text{Y}_{0.25}\text{Yb}_{0.25}\text{Dy}_{0.25}\text{Er}_{0.25})\text{B}_2\text{C}_2$ (HE- REB_2C_2) were investigated.

2 Experimental procedure

2.1 Starting materials

REH_2 (RE = Y, Yb, Dy, and Er) powders with purity of 99.9% and a mean particle size of $\sim 75 \mu\text{m}$ were purchased from Hunan Rare Earth Metal Materials Research Institute Co., Ltd., Changsha, China. B_4C powders with purity of 99% and a mean particle size of $\sim 14 \mu\text{m}$ were purchased from Mudanjiang Diamond Boron Carbide Co., Ltd., Mudanjiang, China. The mean particle size of carbon black (99.9%, EnoMaterial Co., Ltd., Qinhuangdao, China) was $\sim 20 \text{ nm}$. REH_2 , B_4C , and C powders were used as starting materials.

2.2 In-situ reactive SPS process

To fabricate the HE-REB₂C₂ ceramics, REH₂, B₄C, and C powders were mixed in a stoichiometric ratio of 2 : 1 : 3, where the molar ratio of YH₂ : YbH₂ : DyH₂ : ErH₂ in REH₂ was 1 : 1 : 1 : 1. The *in-situ* reactive SPS process was performed in an SPS furnace (HPD-25, FCT Systeme GmbH, Germany) in an Ar atmosphere at a target temperature of 1800 °C for 15 min under a uniaxial pressure of 30 MPa. To promote the *in-situ* reaction, a pre-reaction at 1600 °C for 30 min was set before the target sintering temperature of 1800 °C. The heating and cooling rates were both 50 °C·min⁻¹. For the sake of comparison, the monolithic YB₂C₂, YbB₂C₂, DyB₂C₂, and ErB₂C₂ ceramics were prepared using the same sintering conditions.

2.3 Material characterization

Phase compositions and crystal structures of the samples were identified by an X-ray diffractometer (D8 Advance, Bruker AXS, Germany) equipped with Cu K α radiation ($\lambda = 1.5406 \text{ \AA}$) under operating power of 1600 W (a current of 40 mA and a voltage of 40 kV) at a step scan of $2\theta = 0.02^\circ$ and step time of 0.2 s. The phase compositions and the lattice parameters of the as-synthesized HE-REB₂C₂ were determined by Rietveld refinement of the XRD patterns using the TOPAS-Academic v6 software. Furthermore, α_a and α_c of the as-obtained HE-REB₂C₂ were measured using a high-temperature X-ray diffractometer (D8 Advance, Bruker AXS, Germany) in vacuum at a step scan of $2\theta = 0.02^\circ$ and step time of 0.2 s at the temperature range of 300–773 K. The heating rate was 50 °C·min⁻¹ for 10 min. Expanded lattice parameters were determined by Rietveld analysis of high-temperature X-ray diffraction (XRD) patterns using the TOPAS-Academic v6 software.

Microstructures and chemical compositions of the samples were detected by a scanning electron microscope (SEM; Regulus 8230, Hitachi, Japan) equipped with an energy-dispersive spectroscopy (EDS) system. The mean grain sizes (d) of the samples were determined using several selected SEM images, where at least 100 grains were measured. Thin foils of the samples for transmission electron microscopy (TEM) observations were prepared using the focused ion beam (FIB) technique (Aurgia, Carl Zeiss, USA). Aberration-correction TEM images of HE-REB₂C₂ were observed by a double spherical aberration-correction transmission electron microscope (Spectra

300, ThermoFisher, USA).

The apparent density (ρ) of the samples was measured by Archimedes' method. The specific heat capacity (c_p) and thermal diffusivity coefficient (α) were determined by the laser flash method using a Netzsch equipment (LFA467, NETZSCH- Gerätebau GmbH, Germany) in an argon atmosphere at 323 K. The k (W·m⁻¹·K⁻¹) is calculated by Eq. (1):

$$k = \rho\alpha c_p \quad (1)$$

Vickers hardness of the as-obtained samples was measured by a Vickers hardness tester (Wilson VH3300, Buehler, USA) under loads of 5, 10, 20, and 30 N. The dwell time at the maximum load was 15 s.

3 Results and discussion

3.1 Microstructure of as-obtained HE-REB₂C₂

Figures 1(a)–1(e) show SEM images of the fractured surfaces of REB₂C₂ (RE = Y, Yb, Dy, and Er) and HE-REB₂C₂ samples. A typical nano-laminated structure, similar to MAX phases, was observed for the samples of both REB₂C₂ (RE = Y, Yb, Dy, and Er) and HE-REB₂C₂ samples. The main fracture mechanism observed in all samples was intergranular fracture along with pull-out of the plate-like laminated grains. The laminated fracture, such as kink band, delamination, and slipping, was observed, which suggested ductile nature of both REB₂C₂ (RE = Y, Yb, Dy, and Er) and HE-REB₂C₂ samples. The d of REB₂C₂ (RE = Y, Yb, Dy, and Er) and HE-REB₂C₂, including length and thickness, is shown in Fig. 2. Both the length and the thickness of the HE-REB₂C₂ grains were finer than those of single REB₂C₂ ceramics (RE = Y, Yb, Dy, and Er), due to sluggish diffusion effect of HECs [37]. On the other hand, few pores were observed on the fracture surfaces of all samples, except the sample YbB₂C₂, which showed increased porosity. This was in good agreement with density and open porosity results (Table 1).

Figure 3(a) shows XRD patterns of the REB₂C₂ (RE = Y, Yb, Dy, and Er) and HE-REB₂C₂ samples sintered at 1800 °C. The REB₂C₂ phase was the predominant phase in all samples, while some RE–B (including REB₄ and REB₆), RE–O, and RE–B–O impurities were detected. According to Rietveld refinement results, the amounts of the main YB₂C₂, YbB₂C₂, DyB₂C₂, and ErB₂C₂ phases were 72.91, 60.36, 75.21, and 72.81 wt%, respectively. At the same time, the XRD pattern

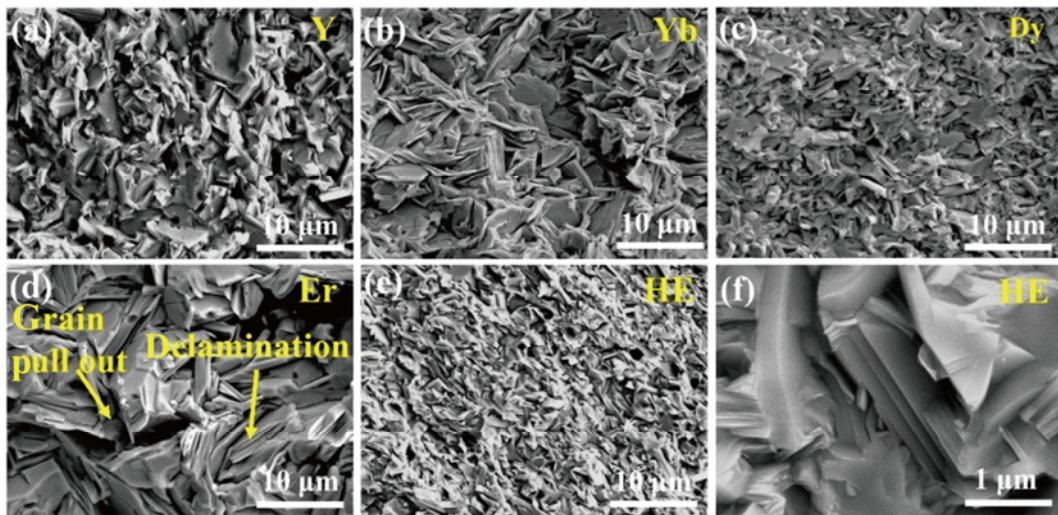


Fig. 1 (a–e) Low- and (f) high-magnification SEM images of fracture surfaces of REB₂C₂ (RE = Y, Yb, Dy, and Er) and HE-REB₂C₂ sintered at 1800 °C.

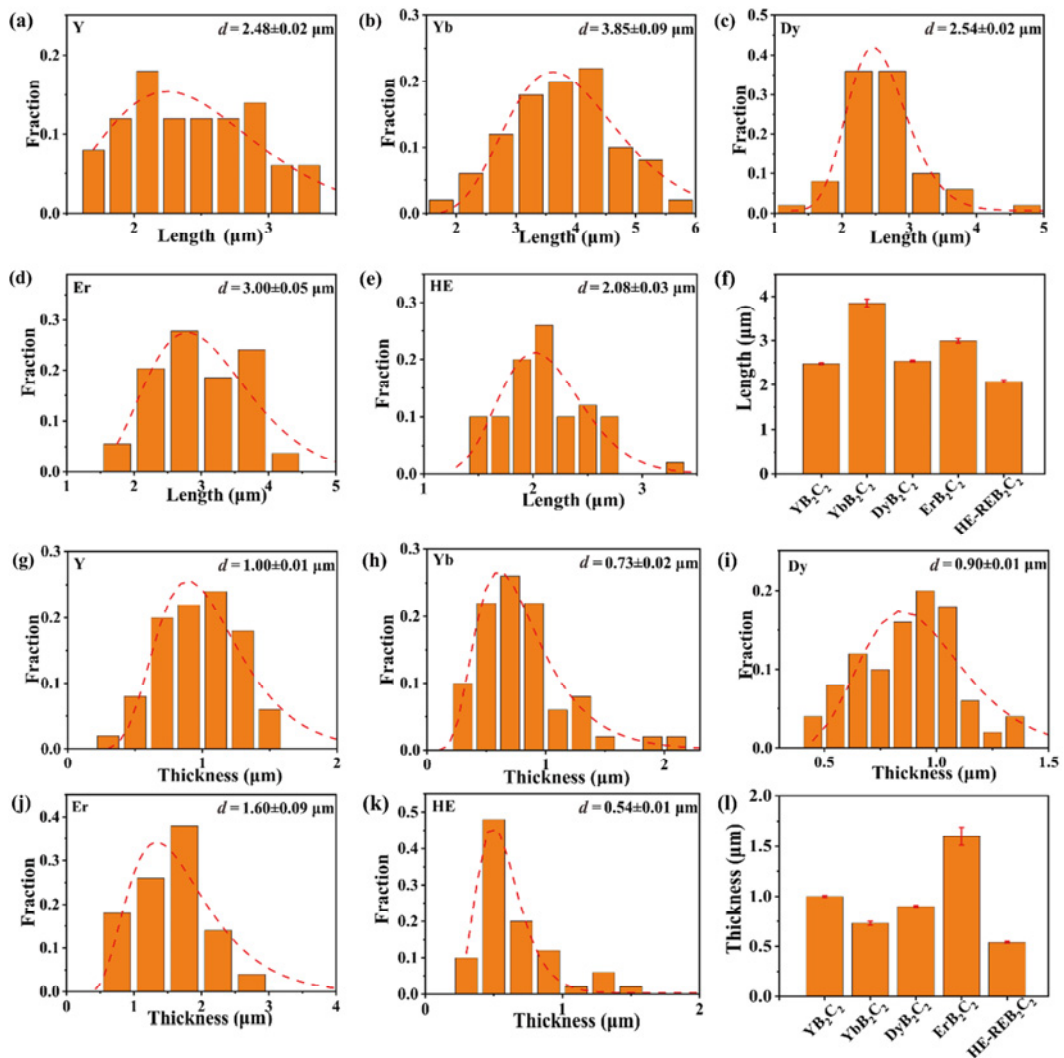
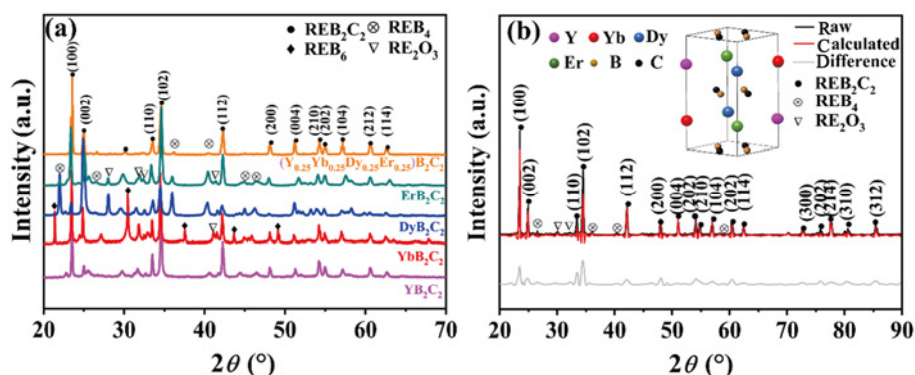


Fig. 2 (a–e) Grain lengths of REB₂C₂ (RE = Y, Yb, Dy, or Er) and HE-REB₂C₂, (f) comparison of grain lengths of REB₂C₂ (RE = Y, Yb, Dy, and Er) and HE-REB₂C₂, (g–k) grain thicknesses of REB₂C₂ (RE = Y, Yb, Dy, and Er) and HE-REB₂C₂, and (l) comparison of grain thicknesses of REB₂C₂ (RE = Y, Yb, Dy, and Er) and HE-REB₂C₂.

Table 1 Open porosity, density, relative density, and grain sizes of high-entropy (Y_{0.25}Yb_{0.25}Dy_{0.25}Er_{0.25})B₂C₂ and REB₂C₂ (RE = Y, Yb, Dy, and Er) samples

Compound	Open porosity (%)	Density (g/cm ³)		Relative density (%)	Grain size (μm)	
		Experimental	Calculated		Length	Thickness
HE-REB ₂ C ₂	0.27	6.12	6.31	97.0	2.08±0.03	0.54±0.01
YB ₂ C ₂	0.17	4.05	4.12	98.3	2.48±0.02	1.00±0.01
YbB ₂ C ₂	1.38	6.56	7.14	91.9	3.85±0.09	0.73±0.02
DyB ₂ C ₂	0.19	6.34	6.49	97.7	2.54±0.02	0.90±0.01
ErB ₂ C ₂	0.13	6.73	6.76	99.6	3.00±0.05	1.60±0.09

**Fig. 3** (a) XRD pattern of as-synthesized (Y_{0.25}Yb_{0.25}Dy_{0.25}Er_{0.25})B₂C₂ and corresponding XRD patterns of YB₂C₂, YbB₂C₂, DyB₂C₂, and ErB₂C₂ and (b) Rietveld refinement of XRD pattern of as-synthesized (Y_{0.25}Yb_{0.25}Dy_{0.25}Er_{0.25})B₂C₂.

of the HE-REB₂C₂ sample was almost identical, which confirmed that both the HE-REB₂C₂ and the REB₂C₂ (RE = Y, Yb, Dy, and Er) materials consisted of the homologous crystal structure. Figure 3(b) presents the typical Rietveld refinement of the XRD pattern of HE-REB₂C₂. Rietveld fitting results (reliability factor (R_{wp}) = 9.863%) showed that a tetragonal structure was the predominant phase, which is a characteristic structure of the REB₂C₂ phase. The sample consisted of 88.63 wt% high-entropy (Y_{0.25}Yb_{0.25}Dy_{0.25}Er_{0.25})B₂C₂ phase with a small amount of REB₄ (8.22 wt%), and RE₂O₃ (3.15 wt%).

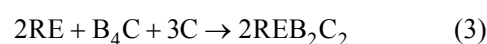
Refined lattice parameters (a and c) of the as-obtained HE-REB₂C₂ and REB₂C₂ (RE = Y, Yb, Dy, and Er) are given in Table 2. The a and c of the HE-REB₂C₂ phase were 3.78 and 7.14 Å, respectively, which were close to the mean values of the corresponding four reference compositions REB₂C₂ (RE = Y, Yb, Dy, and Er). It is interesting to note that the lattice distortion of HE-REB₂C₂ occurred mainly along the c direction, while no significant change was observed for the lattice parameters in the a direction when compared to those of the single-composition REB₂C₂. This can be explained based on the crystal structure of REB₂C₂. In the inner of the B₂C₂ layers and RE layers, there are strong chemical bonds

Table 2 a and c of high-entropy (Y_{0.25}Yb_{0.25}Dy_{0.25}Er_{0.25})B₂C₂ and the reference REB₂C₂ (RE = Y, Yb, Dy, and Er) samples

Compound	Experimental		Theoretical		Reference card
	a (Å)	c (Å)	a (Å)	c (Å)	
HE-REB ₂ C ₂	3.78	7.14	—	—	This study
YB ₂ C ₂	3.78	7.12	3.79	7.12	PDF#27-0971
YbB ₂ C ₂	3.78	7.13	3.78	7.40	ICSD#612680
DyB ₂ C ₂	3.78	7.13	3.78	7.12	PDF#26-0586
ErB ₂ C ₂	3.78	7.02	3.79	7.09	ICSD#612593

between the atoms in the a direction, such as B–C covalent bonds and RE–RE metallic bonds. On the other hand, relatively weak van der Waals forces exist between the B₂C₂ layers and the RE layers along the c direction [19]. Therefore, the lattice distortion of the HE-REB₂C₂ structure was favorable in the c direction.

The possible formation process of HE-REB₂C₂ is similar to the formation mechanism of REB₂C₂ [22]. First, REH₂ decomposed to RE and release H₂. The RE and C diffused to the surface of B₄C to form HE-REB₂C₂. The main reactions can be summarized as Reactions (2)–(4):



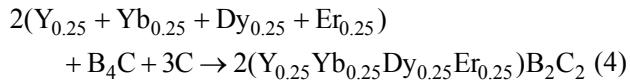


Figure 4 shows an SEM image of the polished surface of the high-entropy $(\text{Y}_{0.25}\text{Yb}_{0.25}\text{Dy}_{0.25}\text{Er}_{0.25})\text{B}_2\text{C}_2$ ceramics and the corresponding rare-earth elemental distributions of Y, Yb, Dy, and Er. The EDS mappings indicated that the four rare-earth elements of Y, Yb, Dy, and Er were homogeneously distributed in the selected area at the micrometer scale, confirming chemical homogeneity of the HE-REB₂C₂ sample. Furthermore, the semi-quantitative EDS analysis results of the spots 1–3 are given in Table 3. The results showed that the atomic ratios of the rare-earth elements (Y, Yb, Dy, and Er) are approximately equal, which indicated that the multiple rare-earth elements had similar solid solubility in the as-obtained HE-REB₂C₂ ceramics.

To further confirm the microstructures and phase compositions of the HE-REB₂C₂ samples, the atomic-scale microstructure along the [001] zone axis was performed by high-resolution transmission electron microscopy (HR-TEM) and corresponding selected area electron diffraction (SAED) pattern, as shown in Fig. 5. It revealed that the as-synthesized HE-REB₂C₂ possessed a single crystal tetragonal structure as a result of the well-arranged diffraction spots with the symmetry. Figures 5(b) and 5(c) show the atomic-resolution high-angle annular dark field (HAADF) scanning transmission electron microscopy (STEM) image and annular bright field (ABF) STEM image of the HE-REB₂C₂ ceramics, respectively. The atomic-resolution HAADF and ABF images showed an

atomic configuration of atoms in the view of the [001] zone axis. The lattice fringe spacing of 0.3956 nm were assigned to the (100) plane of HE-REB₂C₂. The HAADF image is related to the *z* contrast, which is proportional to the square of an atomic number [38]. The contrast of ABF image is proportional to 1/3 of an atomic number, so it is more sensitive to light elements than the HAADF image [39]. Therefore, it clearly showed the positions of the light atoms, such as B and C. The atomic configuration of the atoms (Figs. 5(b) and 5(c)) indicated that the planes of rare-earth atom and the planes of B–C atom planes are alternately stacked, which is consistent with the crystal structure of HE-REB₂C₂ detected by XRD.

3.2 CTEs of HE-REB₂C₂

Thermal expansion behavior is a critical thermal property of the materials for ultra-high-temperature applications, which is caused by anharmonic vibration of atoms in a temperature field [40,41]. Therefore, the expansion of the lattice parameters of the as-obtained HE-REB₂C₂ was detected using high-temperature XRD at the temperature range of 303–773 K, as shown in Fig. 6. There was no phase decomposition and transformation in the measured temperature range, which implied that the HE-REB₂C₂ material was stable at the temperatures up to 773 K. The evolution of the (100) and (002) peaks with the increasing temperature indicates the thermal expansion of the HE-REB₂C₂ lattice along the *a* and *c* directions, respectively. It was clearly shown that no significant shift of the (100) peak was

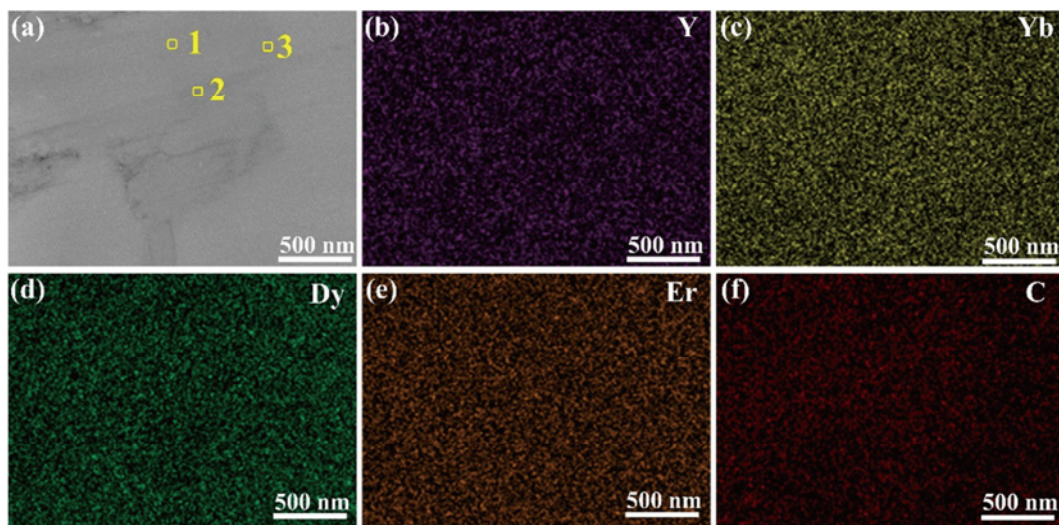


Fig. 4 (a) SEM image of polished surface of high-entropy $(\text{Y}_{0.25}\text{Yb}_{0.25}\text{Dy}_{0.25}\text{Er}_{0.25})\text{B}_2\text{C}_2$ ceramics and (b–f) corresponding EDS elemental mappings.

Table 3 EDS results of spots 1–3 in Fig. 4(a)

No.	Atomic composition (at%)			
	Y	Yb	Dy	Er
1	16.8	27.9	27.0	28.3
2	15.2	25.9	28.2	30.7
3	18.1	28.9	29.2	23.8

observed with the increase in the temperature (Fig. 6(b)), while the (002) peak was obviously shifted to the left with the increasing temperature. This clearly reveals the anisotropy of the thermal expansion of the HE-REB₂C₂ structure in the *a* and *c* directions. The expanded lattice parameters were calculated by Rietveld refinements using the TOPAS-Academic v6 software.

The α_a and α_c can be calculated by Eqs. (5) and (6), respectively [42]:

$$\alpha_a = \frac{d(a(T))}{a_0 dT} \quad (5)$$

$$\alpha_c = \frac{d(c(T))}{c_0 dT} \quad (6)$$

where $\frac{d(a(T))}{dT}$ and $\frac{d(c(T))}{dT}$ are the slopes of the fitted linear functions of $a(T)$ and $c(T)$, respectively, and a_0 and c_0 are the lattice parameters at room temperature (~303 K). Furthermore, the average CTE

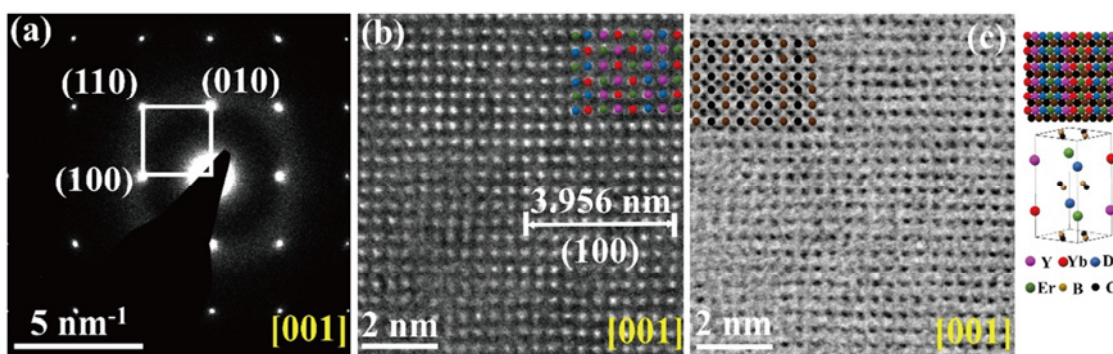


Fig. 5 (a) SAED pattern of HE-REB₂C₂ sample, (b) atomic-resolved HAADF image of HE-REB₂C₂ showing atomic arrangements along [001] direction, and (c) atomic-resolved ABF image of HE-REB₂C₂.

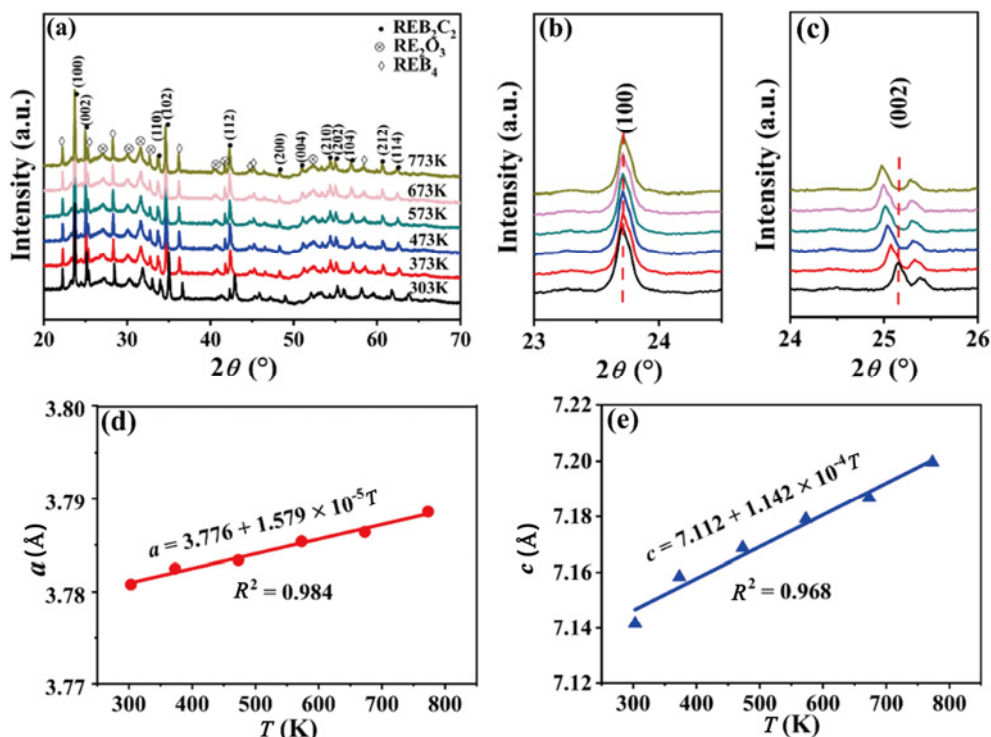


Fig. 6 (a) XRD patterns of high-entropy (Y_{0.25}Yb_{0.25}Dy_{0.25}Er_{0.25})B₂C₂ structure at different temperatures, (b, c) partial XRD patterns of high-entropy (Y_{0.25}Yb_{0.25}Dy_{0.25}Er_{0.25})B₂C₂ showing peak shift of (100) and (002) planes, respectively, and fitted lattice parameters of (d) *a* and (e) *c* as a function of *T* (R^2 reflects the fitness of the curve to experimental values).

(α_{av}) can be obtained by Eq. (7) [42]:

$$\alpha_{av} = \frac{2\alpha_a + \alpha_c}{3} \quad (7)$$

According to the calculated results of Eqs. (5)–(7), α_a , α_c , and α_{av} of the as-obtained HE-REB₂C₂ were 4.18, 16.06, and 8.14 μK^{-1} , respectively. This clearly confirmed that the HE-REB₂C₂ structure exhibits the thermal expansion anisotropy along the *a* and *c* directions. The α_c/α_a ratio is usually used to characterize the degree of the anisotropy of the thermal expansion. The ratio of α_c/α_a calculated for the HE-REB₂C₂ structure was 3.84. It should be noted that the value of α_c/α_a ratio of HE-REB₂C₂ is significantly greater than those of some typical MAX phases, such as Ti₃SiC₂ (1.09) [43], Ti₃AlC₂ (1.33) [44], and Ti₂AlC (1.41) [45]. This suggested the anisotropy in the thermal expansion of HE-REB₂C₂ is more significant when compared to the MAX phase. This phenomenon can be explained by the crystal structure of HE-REB₂C₂. In REB₂C₂, the RE and B₂C₂ layers are alternatively stacked along the *c* direction to form a layered structure [17]. The chemical bonds between the RE and B₂C₂ layers are RE–B and RE–C bonds along the *c* direction, which are weaker than the highly covalent B–C bonds in the B₂C₂ planes along the *a* direction. According to the calculated results of the YB₂C₂ structure, the length of B–C bonds within the four-member B–C rings is 1.6196 Å, while the lengths of C–C and B–B bonds that connect the eight-member B–C rings are 1.2755 and 1.7613 Å, respectively. On the other hand, the lengths of Y–C and Y–B bonds are 2.6789 and 2.7477 Å, respectively [19]. Such a significant difference in bond lengths (roughly inferring the bond energy) along the *a* and *c* directions resulted in the obvious anisotropic thermal properties of the HE-REB₂C₂ structure.

3.3 Thermal conductivity of HE-REB₂C₂ and REB₂C₂

Figure 7 shows thermal conductivity of the as-obtained HE-REB₂C₂ and REB₂C₂ (RE = Y, Yb, Dy, and Er) materials. It should be noted that, the lowest thermal conductivity was measured for the YbB₂C₂ sample, due to its highest porosity (lowest relative density), as shown in Table 1. Since the result was strongly affected by the porosity of the sample, this sample was not compared to the HE-REB₂C₂ material.

The thermal conductivity of the HE-REB₂C₂ sample was $9.2 \pm 0.09 \text{ W} \cdot \text{m}^{-1} \cdot \text{K}^{-1}$, which was lower than those of YB₂C₂ ($19.2 \pm 0.07 \text{ W} \cdot \text{m}^{-1} \cdot \text{K}^{-1}$), DyB₂C₂ (11.9 ± 0.06

$\text{W} \cdot \text{m}^{-1} \cdot \text{K}^{-1}$), and ErB₂C₂ ($12.1 \pm 0.03 \text{ W} \cdot \text{m}^{-1} \cdot \text{K}^{-1}$). This can be attributed to the high-entropy effect. The presence of four rare-earth elements in the solid solution leads to lattice distortion, which improves phonon scattering and increases thermal resistance [46]. On the other hand, the impurities of RE–B and RE–O, which have higher thermal conductivity than that of HE-REB₂C₂ [47,48], also affect the thermal conductivity. Therefore, the higher purity of the high-entropy (Y_{0.25}Yb_{0.25}Dy_{0.25}Er_{0.25})B₂C₂ phase (88.63 wt%) resulted in lower thermal conductivity when compared to those of the reference single-composition REB₂C₂ (RE = Y, Dy, and Er). Furthermore, the grain sizes of the HE-REB₂C₂ material (both the length and the thickness of the grains) were lower than those of YB₂C₂, DyB₂C₂, and ErB₂C₂ due to the sluggish diffusion effect of HECs. The phonon scattering caused by grain boundaries increases as the grain size decreases. Therefore, the sample with the finest grain size (HE-REB₂C₂) contained the highest number of grain boundaries, at which the phonon scattering occurs, resulting in the lowest thermal conductivity among all of the investigated materials [49]. In addition, it should be noted that the electronic thermal conductivity of HE-REB₂C₂ was also changed when compared to those of the reference single-composition REB₂C₂ (RE = Y, Yb, Dy, and Er) due to the high-entropy effect. The contribution of the electric thermal conductivity of REB₂C₂ and HE-REB₂C₂ is unclear until now, and we will investigate it in the future work.

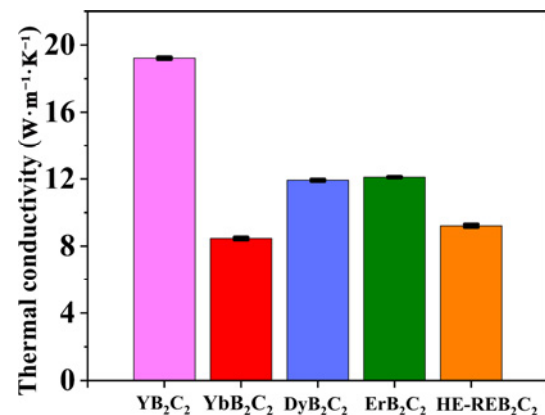


Fig. 7 Thermal conductivity of as-obtained HE-REB₂C₂ and REB₂C₂ (RE = Y, Yb, Dy, and Er) materials.

3.4 Vickers hardness of HE-REB₂C₂ and REB₂C₂

Figure 8(a) shows Vickers hardness as a function of the load for the as-obtained high-entropy (Y_{0.25}Yb_{0.25}Dy_{0.25}Er_{0.25})B₂C₂ and REB₂C₂ (RE = Y, Yb, Dy, and Er)

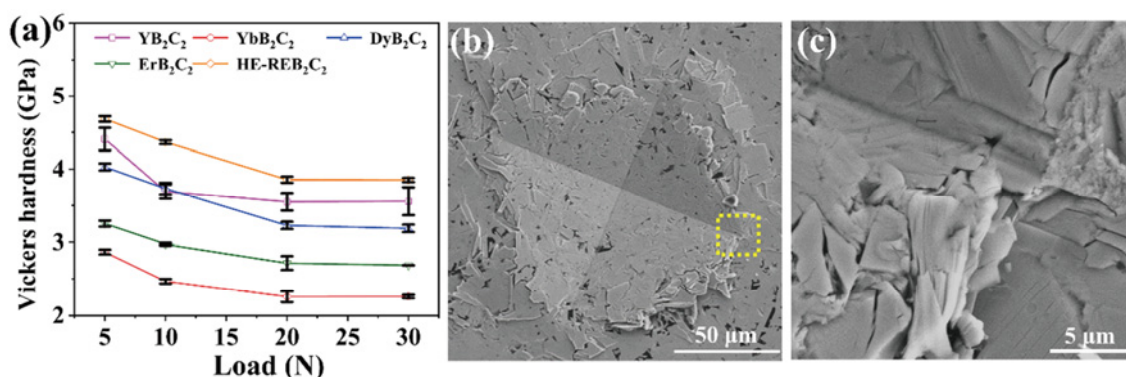


Fig. 8 (a) Vickers hardness of high-entropy ($Y_{0.25}Yb_{0.25}Dy_{0.25}Er_{0.25}$)B₂C₂ and REB₂C₂ (RE = Y, Yb, Dy, and Er) materials as a function of load and (b, c) low- and high-magnification SEM images of HE-REB₂C₂ surface after Vickers indentation under load of 20 N, respectively.

materials as a function of load. The hardness of the HE-REB₂C₂ material gradually decreased from 4.68 ± 0.04 to 3.85 ± 0.03 GPa as the load increased from 5 to 30 N, respectively, due to indentation load size effect [50]. On the other hand, Vickers hardness of HE-REB₂C₂ samples was slightly higher than the hardness of the reference single-composition REB₂C₂ (RE = Y, Yb, Dy, and Er). For example, Vickers hardness of HE-REB₂C₂ and REB₂C₂ (RE = Y, Yb, Dy, and Er) measured under the load of 30 N were 3.85 ± 0.03 , 3.56 ± 0.19 , 2.26 ± 0.02 , 3.19 ± 0.05 , and 2.69 ± 0.03 GPa, respectively. This can be attributed to the disorder among rare-earth species and the solid solution hardening mechanism [51]. The multi rare-earth element doping led to the increased configuration entropy and the intensified lattice distortion, which increased the resistance against dislocation movement. As a result, the hardness of HE-REB₂C₂ was improved [52]. It is interesting that the hardness of the HE-REB₂C₂ was just slightly higher than the hardness of the reference YB₂C₂. It can be ascribed to the fact that the weak chemical bonds between the RE and B₂C₂ layers along the *c* direction, which led to slipping and delamination preferentially preference along the basal plane, as shown in Figs. 8(b) and 8(c). As a result, the solid solution hardening effect was not so much obviously, as was reported in some typical high-entropy borides and carbides [53,54].

Figures 8(b) and 8(c) show low- and high-magnification SEM images of the HE-REB₂C₂ surface after Vickers indentation under a load of 20 N, respectively. No typically extended cracks propagating from corners of the indents, which are usually observed for brittle ceramics, were found in this study, suggesting ductile nature of HE-REB₂C₂ ceramics [55]. The morphology of Vickers indents was irregular with

exfoliated surfaces and deformed particles, and was similar to the shapes of the indent in the MAX phases [56]. When the basal plane was parallel to the load, the laminated fracture was observed owing to the slipping, delamination, and kink formation. On the other hand, the exfoliation occurred when the basal plane was perpendicular to the applied load. The deformation during the indentation of HE-REB₂C₂ can be ascribed to the basal plane slipping and debonding at the interface between the RE and B₂C₂ layers [19].

4 Conclusions

In summary, the novel high-entropy rare-earth metal diborodiboride ($Y_{0.25}Yb_{0.25}Dy_{0.25}Er_{0.25}$)B₂C₂ was successfully fabricated using an *in-situ* solid state reaction via SPS. The as-obtained HE-REB₂C₂ showed the single solid solution with a typical tetragonal structure. The four rare-earth elements, such as Y, Yb, Dy, and Er, occupied the RE position in the REB₂C₂ structure, which was revealed by the atomic-resolved HAADF and ABF images of HE-REB₂C₂. The evolution of lattice parameters of HE-REB₂C₂ was determined by the high-temperature XRD, followed by Rietveld refinements. The thermal expansion anisotropy of HE-REB₂C₂ along the *a* and *c* directions was observed. The value of α_c/α_a was 3.84, confirming the significant anisotropy of the HE-REB₂C₂ structure. Compared to the reference REB₂C₂ compositions (RE = Y, Dy, and Er), HE-REB₂C₂ exhibited lower thermal conductivity (9.2 ± 0.09 W·m⁻¹·K⁻¹) due to the high-entropy effect and sluggish diffusion effect. In addition, Vickers hardness of the HE-REB₂C₂ material was 3.85 ± 0.03 GPa, which was just slightly higher than that of the reference YB₂C₂. The investigated

novel ductile nature of the HE-REB₂C₂ ceramics paved the way for the applications as the new class of UHTCs in the aerospace area.

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Declaration of competing interest

The authors have no competing interests to declare that are relevant to the content of this article.

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